

# Introduction to X-ray Absorption Spectroscopy Steve Heald, X-ray Science Division, APS

Part 1:

- •What is XAFS?
- •What does it tell us?
- The EXAFS equation

Part 2:

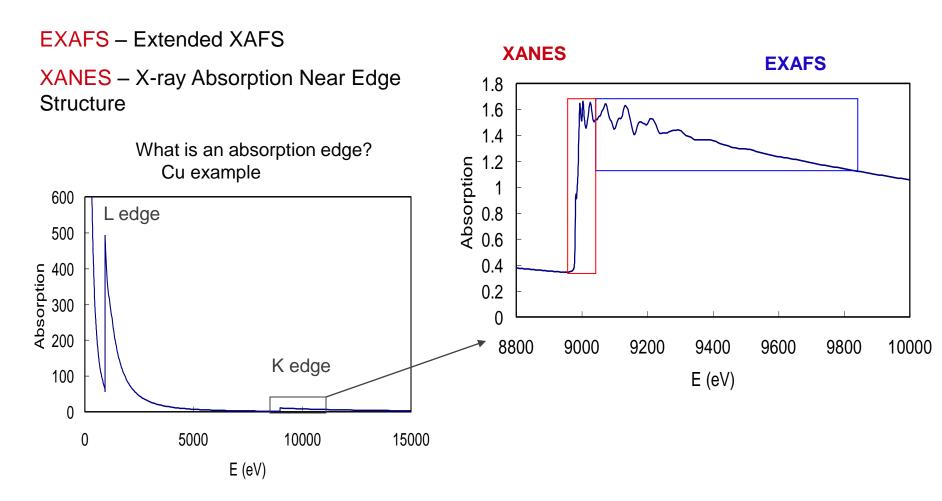
•Basic steps in the analysis

- •Quick overview of typical analysis
- •Example for Co doped ZnO films



## What is XAFS?

XAFS stands for X-ray Absorption Fine Structure and appears in the vicinity of absorption edges



### EXAFS sensitive to local atomic structure

Absorption (arb. units)

Nearest neighbor information:

Distance

Atomic type

Coordination number

Vibrational amplitude

Typically useful for the first few coordination shells

Kr under pressure 8.0 4.0 2.0 GPa 1.15 GPa 0.75 GPa 0.60 GPa 0.35 GPa 0.1 GPa gas Kr T = 300 K14325 14350 14375 14400 E(eV)

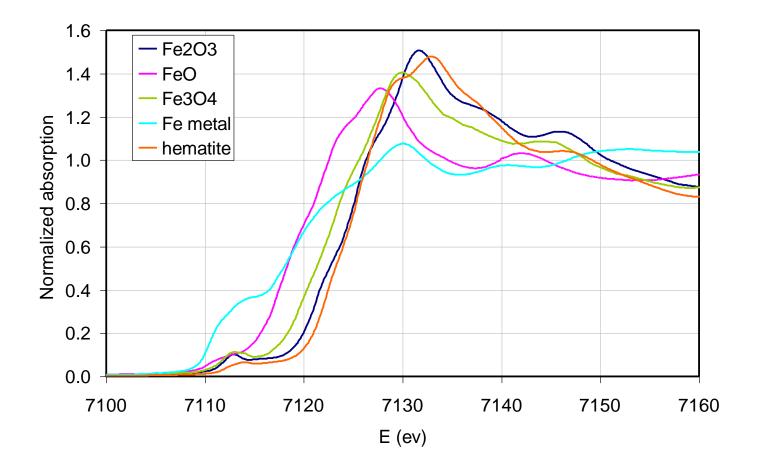
GPa

GPa

DiCicco etal PRB 54, 9086

### XANES sensitive to chemical state and valence

Fe compounds



# Why is x-ray absorption useful?

- Short range order probe crystallinity not required
- Element specific (Fe 7111 eV, Cu 8980, etc.)
- Near edge sensitive to chemistry
  - Valence
  - Site symmetry
- Extended structure sensitive to atomic neighbors
  - Number and type of neighbors
  - Distance to neighbors
  - Disorder in neighbors
  - Orientation of bonds (using x-ray polarization)
- Can be applied to dilute systems

# **Calculation of Absorption**

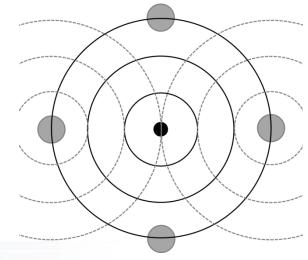
From Fermi's golden rule

$$\mu = \frac{4\pi\omega e^2}{c} N \left| \left\langle i \left| r \right| f \right\rangle \right|^2 \rho(E_f)$$

i – Initial state: 1s or 2p electron localized in the core of the atom

f – Final state: outgoing photoelectron wave + interactions with the neighboring atoms

 $\rho(E_f)$  – Density of final states



# The XAFS equation

$$\chi(k) = \frac{\mu(k) - \mu_0(k)}{\mu_0(k)} = \sum_j \frac{N_j}{kR_j^2} A_j(k) p_j(\mathbf{e}) \sin\left[2kR_j + \psi_j(k)\right]$$

 $k = \frac{\sqrt{2m_e E}}{\hbar}$  The electron wavevector,  $E = E_{x-ray} - E_{edge}$ 

 $\mu_0$  - "atomic" absorption

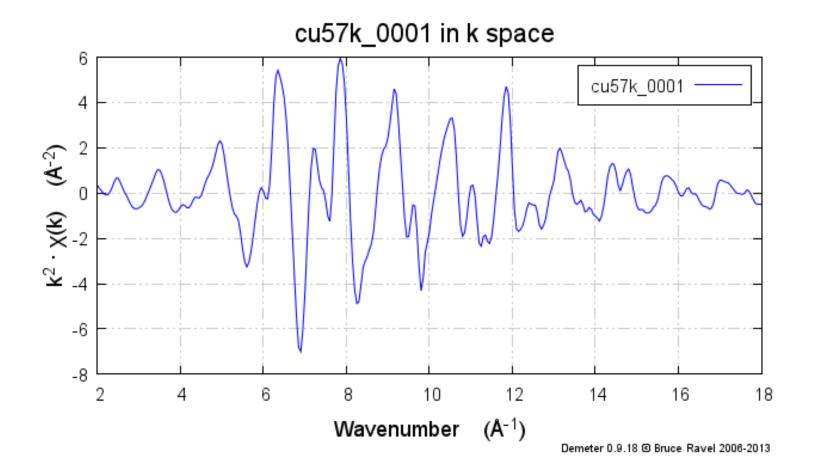
R - distance to atoms of shell j

N - number of atoms in shell j

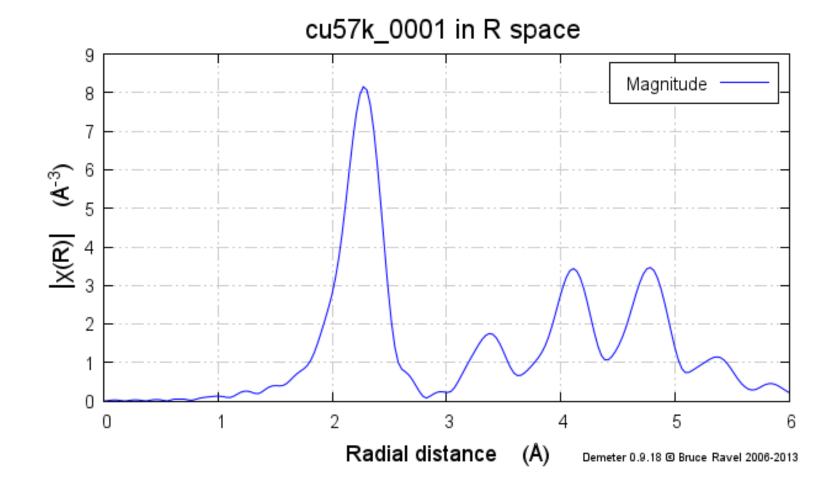
A - amplitude factor

- p polarization factor
- $\psi$  phase factor

### **Cu EXAFS**



### Fourier Transform can separate the shells



### The amplitude factor

$$\chi(k) = \sum_{j} \frac{N_{j}}{kR_{j}^{2}} A_{j}(k) p_{j}(e) \sin\left[2kR_{j} + \psi_{j}(k)\right]$$

 $A_{j}(k) = S_{0}^{2}F_{j}(k)Q_{j}(k)e^{-2R_{j}/\lambda}$ 

 $S_0^2$  - amplitude reduction due to multi-electron excitation Typically 0.7 -0.9 (almost constant with *k*)

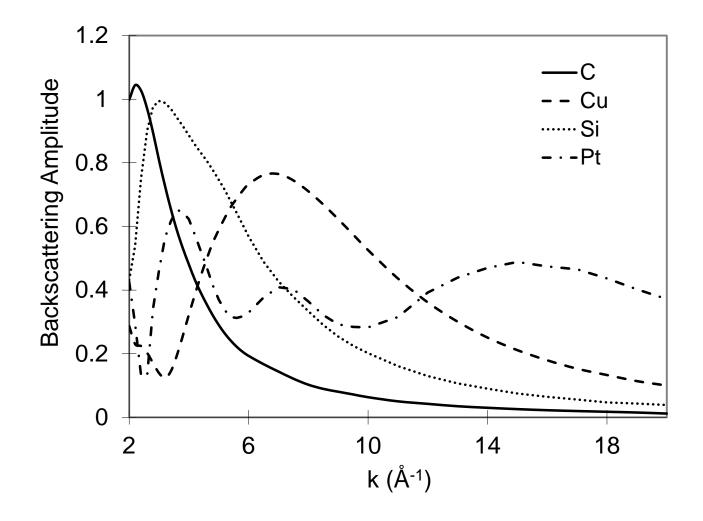
 $F_{j}(k)$  - magnitude of the complex backscattering amplitude  $f_{j}(k,\pi)$ 

Q<sub>i</sub>(k) - disorder term

 $\lambda$  - mean free path of photoelectron (includes contribution from core hole lifetime)

 $f_j(k,\pi)$  calculated with theory such as FEFF





### The Phase factor

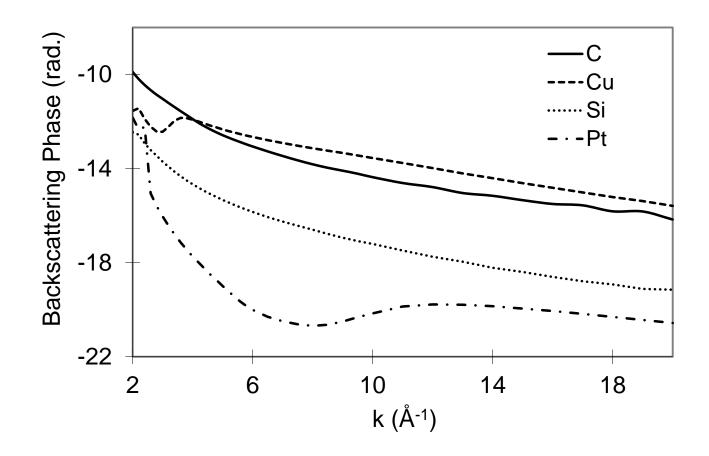
$$\chi(k) = \sum_{j} \frac{N_{j}}{kR_{j}^{2}} A_{j}(k) p_{j}(\mathbf{e}) \sin\left[2kR_{j} + \psi_{j}(\mathbf{k})\right]$$

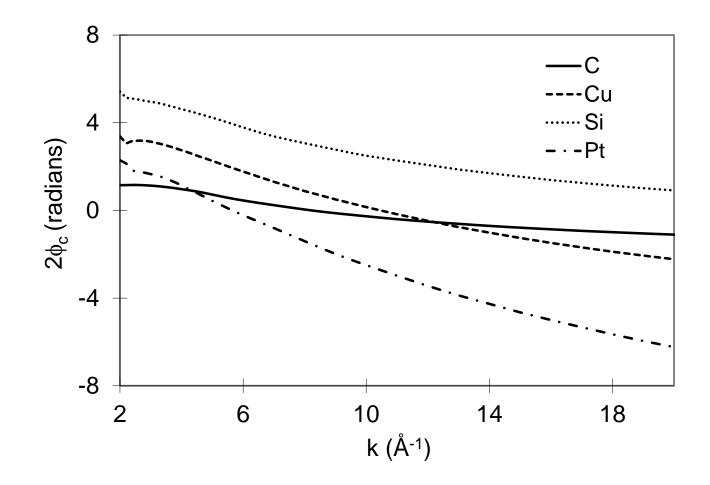
 $\psi_j(k) = 2\phi_c + \theta_j(k) + \varphi_j(k)$ 

 $\phi_{c}$  - p-wave (for K-edge) phase shift due to central atom potential

 $\theta_i(k)$  - the phase of the backscattering factor  $f_i(k,\pi)$ 

 $\phi_i(k)$  - phase factor related to the disorder of the j<sup>th</sup> shell





### **Polarization Factor**

$$\chi(k) = \sum_{j} \frac{N_{j}}{kR_{j}^{2}} A_{j}(k) p_{j}(\mathbf{e}) \sin\left[2kR_{j} + \psi_{j}(k)\right]$$

Synchrotron radiation generally linearly polarized

For the K-shell the photoelectron emitted with p-wave symmetry:

3-fold or higher symmetry  $\langle \cos^2\theta \rangle = 1/3$ , p=1

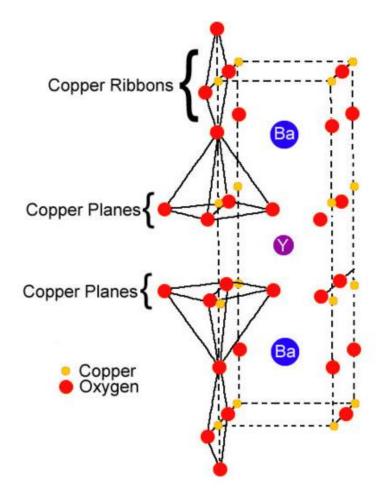
Uniaxial symmetry (i.e. surface):

 $\chi(\Theta) = [2\chi_0 \cos^2(\Theta) + \chi_{90} \sin^2(\Theta)]/3$ 



#### YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub> structure

Sample is oriented powder: c-axis of all grains are aligned, no alignment of a-b axes.



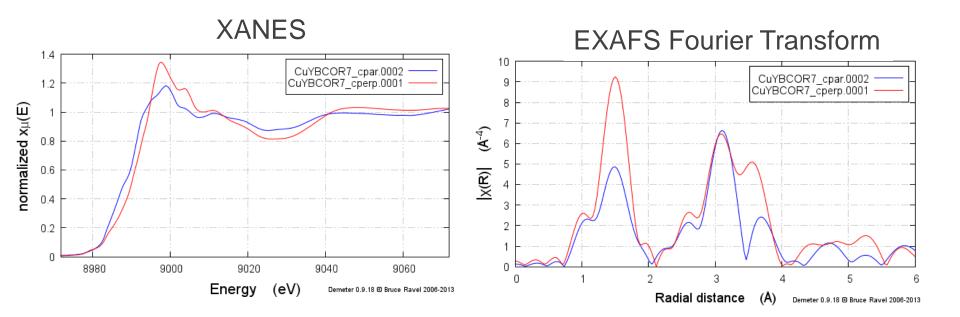
Two types of Cu sites: Planes – have bonds in a-b plane Ribbons – have bonds along c axis

Consider the Cu-O near neighbor bonds:

Polarization in a-b plane both sites contribute, but dominated by Plane sites

Polarization along c axis the signal is dominated by the Ribbon sites

### Polarization dependence of YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub>



### **Disorder terms**

$$\psi_{j}(k) = 2\phi_{c} + \theta_{j}(k) + \phi_{j}(k)$$
$$A_{j}(k) = S_{0}^{2}F_{j}(k) Q_{j}(k) e^{-2R_{j}/\lambda}$$

$$Q_j(k) \exp\left[i\varphi_j(k)\right] = \int dr P_j(r) \exp\left[i2k(r-R_j)\right]$$

 $P_{i}(r)$  - real space distribution of atoms in shell j

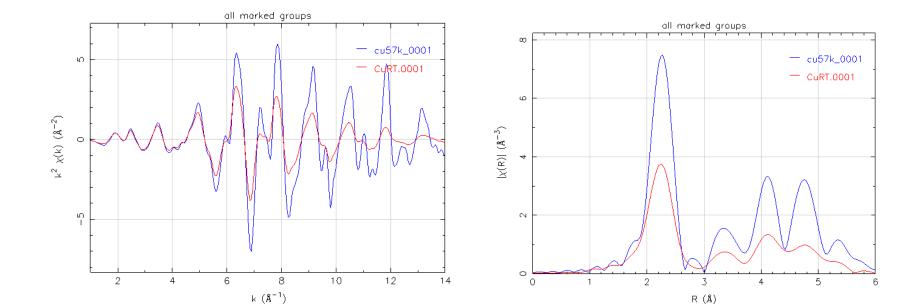
Simplest case is Guassian disorder (i.e. thermal motion) To lowest order:

 $Q_j(k) = \exp(-2k^2\sigma_j^2)$  (Debye-Waller factor)  $\varphi_j(k)=0$ 

 $\sigma_j^2$  - mean squared vibrational amplitude of bond length Different from the diffraction DW factor

Non-symmetric distributions can be handled with the cumulant method

# Change in disorder with temperature



# Choice of Absorption edge

- Low energies require high vacuum and x-ray penetration small
- At high energies lifetime broadening washes out the spectra
- Most convenient to work in range 4-30 keV

K edges:

Ca →Sn

L edges:

#### $\mathsf{Sn} \to \mathsf{U}$

For lighter elements must work at low energies

# L edges

L<sub>1</sub> - 2s initial state - Same symmetry as K edge

 $L_{2,3}$  - 2p initial state - can have transitions to s or d final states

Three possible contributions to  $\chi(k)$ :  $\chi_{ss}$ ,  $\chi_{sd}$ ,  $\chi_{dd}$ 

$$\chi = \left(M_{21}^2 \chi_{dd} + M_{01}^2 \chi_{ss} + 2M_{01}M_{21}\chi_{sd}\right) / \left(M_{21}^2 + M_{01}^2/2\right)$$

Each contribution has its own polarization dependence

$$p_{22}^{j} = \frac{1}{2} \left( 1 + 3 \left\langle \cos^{2} \theta \right\rangle_{j} \right)$$
$$p_{00}^{j} = \frac{1}{2}$$
$$p_{02}^{j} = \frac{1}{2} \left( 1 - 3 \left\langle \cos^{2} \theta \right\rangle_{j} \right)$$



# L edges (cont.)

Two important simplifications:

M<sub>01</sub> typically about 0.2M<sub>21</sub>

 $\rightarrow$  can generally ignore  $M_{01}^2$  term

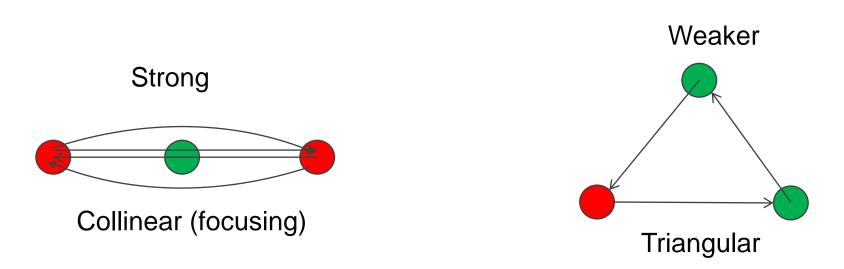
unoriented or symmetric samples:  $p_{02} = 0$ 

For unoriented or symmetric samples L edges can be treated the same as the K edge

For oriented asymmetric samples (i.e. surface site) the crossterm must be included (40% of dd term)



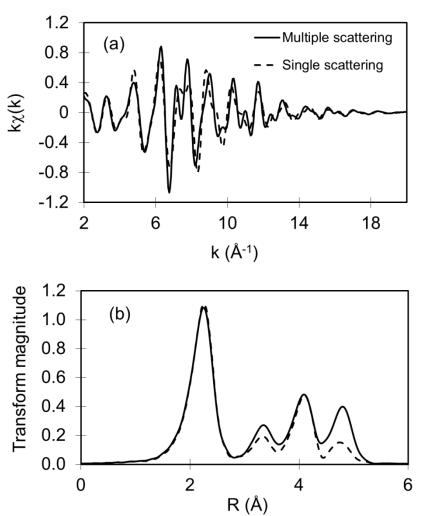
# **Multiple Scattering**



- Not important for first shell
- Must be considered in analyzing all shells above first
- Dominates at large distances
- Typically limits analysis to first few shells



# **Multiple scattering in Cu**



R	Ν	Rel. Amp	Туре
2.556	12	100	Single
3.615	6	23	Single
3.834	48	10	Triangle
4.364	48	9	Triangle
4.428	24	55	Single
4.770	48	11	Triangle
4.770	96	22	Triangle
5.112	12	19	Single
5.112	24	44	Collinear
5.112	12	8	Collinear
5.112	12	33	Collinear

# Measuring XAFS

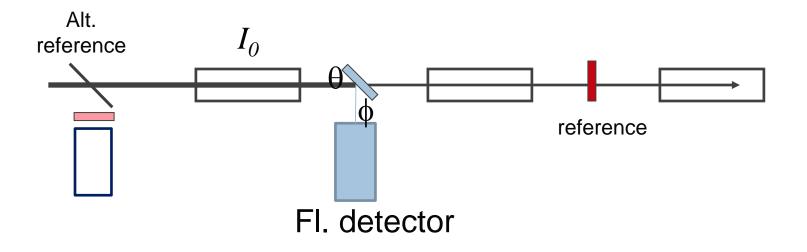
Need to measure absorption

Direct method:

 $I = I_0 e^{-\mu t}$  $\mu t = -\ln(I/I_0) = \ln(I_0/I)$ 

Indirect methods (proportional to absorption) Fluorescence Electron Yield Reflectivity Optical emission

## **Fluorescence Detection**



Need to consider absorption going in and out of the sample

 $\mu'(E) = \frac{I_f}{I_0} = \frac{\mu(E)\sin(\theta)}{\mu_t(E)/\sin\theta + \mu_t(E_f)/\sin\varphi}$ 

 $\mu_t$  - total absorption of sample ( $\mu + \mu_{matrix}$ )



### Fluorescence detection - cont.

Typical case  $\sin\theta = \sin\phi = 45$ 

**Dilute sample**:  $\mu << \mu_{matrix}$   $\frac{I_f}{I_0} = \frac{\mu(E)}{\mu_{matrix}(E) + \mu_{matrix}(E_f)}$  Denominator almost constant

**Concentrated sample**:  $\mu = \mu_{\text{matrix}}$   $\frac{I_f}{I_0} = \frac{\mu(E)}{\mu(E) + \mu(E_f)}$  For K edge:  $\mu(E_f) \approx \mu(E)/8$ (self absorption)

**Thin concentrated sample**: absorption of x-rays entering and exiting sample are negligible

$$\frac{l_f}{l_0} \approx \mu(E)$$

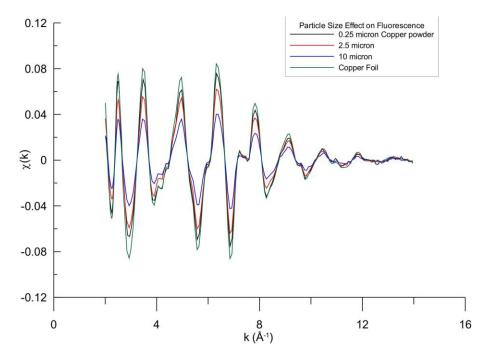
**Glancing exit angle**:  $sin\phi \ll 1$ 

$$\frac{I_f}{I_0} \propto \frac{\mu(E)}{\mu_{matrix}(E_f)}$$



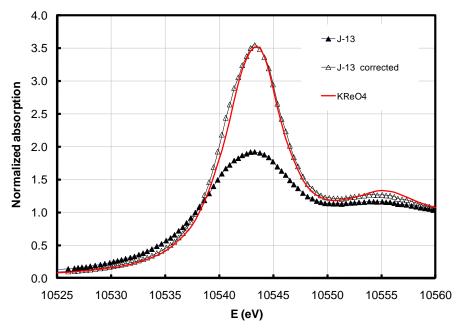
### Self absorption

#### Cu metal



Data from R. Gordon

#### Re near edge



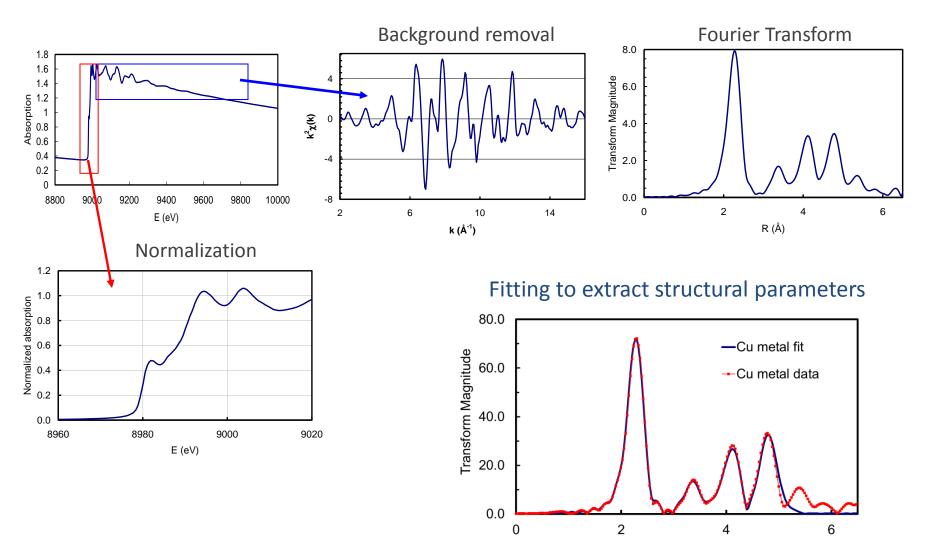
### Fluorescence or transmission?

From statistics –  $\Delta \mu x < 0.05 - 0.1$  fluorescence generally better

Many factors can influence this monochromator glitches sample uniformity other fluorescence lines

Often can measure both simultaneously and choose the best one

# Part 2: Analyzing the EXAFS



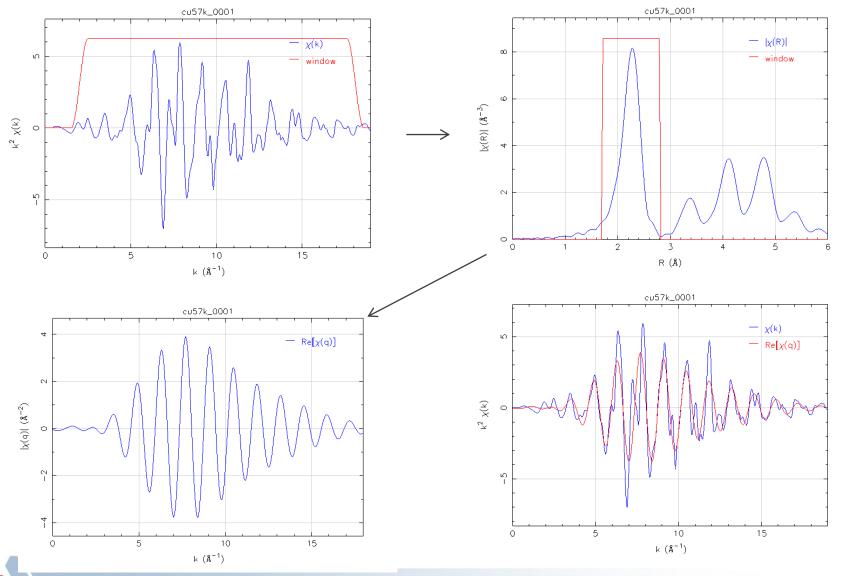
0

R (Å)

4

6

# **Fourier filtering**



# **Independent Data Points**

From information theory:

 $N_{I} = 2\Delta k\Delta R/\pi + 1$ 

 $\Delta \mathbf{k}$  - range of data used in k-space

 $\Delta R$  - range of data used in R-space

Example:

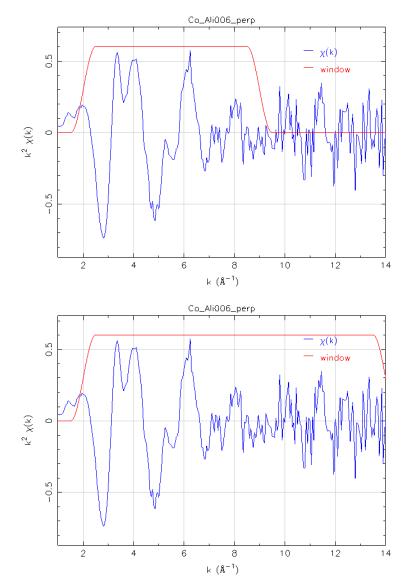
k range is 2 -18 Å<sup>-1</sup>

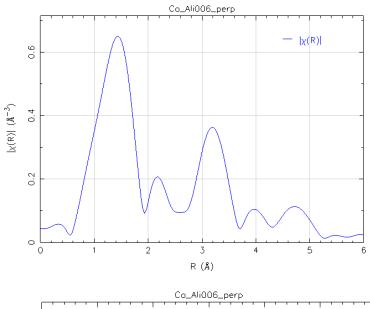
R space filtering or fitting window is 1.7 - 2.8 Å Then maximum number of independent fitting

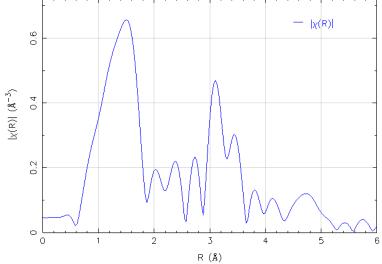
parameters is 11

If data range short (kmax  $\sim$  10) then number of possible fitting parameters will be constrained.

# Example of noisy data







# Typical fitting parameters

- $\Delta E_0$  match the experimental and theoretical edge positions
- Overall amplitude factor (S<sub>0</sub><sup>2</sup>)
- Each shell needs disorder factor (σ<sup>2</sup>)
- May also need ΔR

 $\Delta E_0$  and  $S_0^2$  can generally be common to all shells

Therefore, typically need 4 parameters for simple first shell and 2-3 more for each additional shell.

Always good to constrain the fits with other knowledge about the sample.

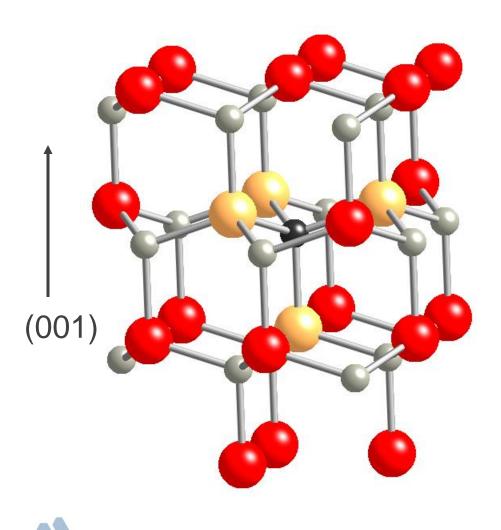
# Co doped ZnO

- Candidate material for a dilute magnetic semiconductor
- Many reports of magnetism at room temperature
- Some cases have been shown to be due to secondary phases (metallic nanoparticles)
- Some cases of well characterized single phase materials without magnetism

#### **EXAFS and XANES are sensitive to secondary phases**

Element specific - allows detection of dilute dopants
XANES can see valence change if metallic Co present
EXAFS can identify the secondary phase (dilute nanoparticles difficult to see with diffraction)
Bulk probe of whole film

# ZnO structure (wurtzite)

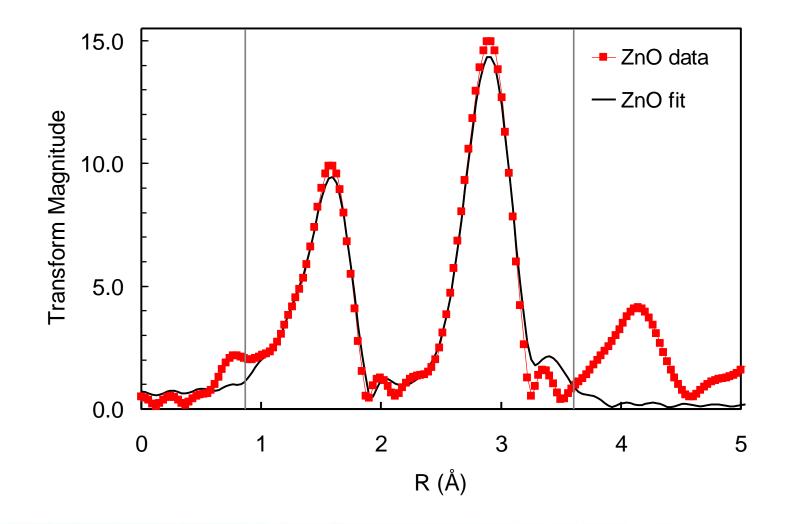


First shell tetrahedrally coordinated - no orientation dependence

Higher shells have more orientation dependence

2 substrate orientations: r-plane sapphire – film grows (110) c-plane sapphire – film grows (001)

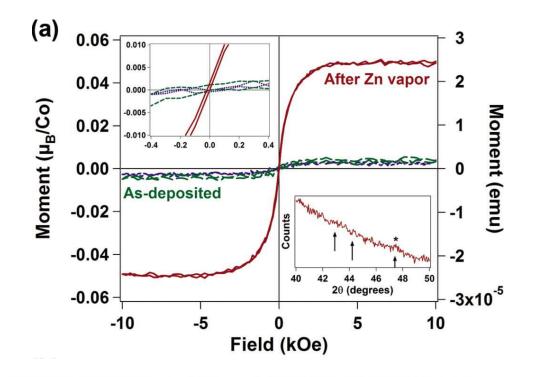
#### **ZnO EXAFS**



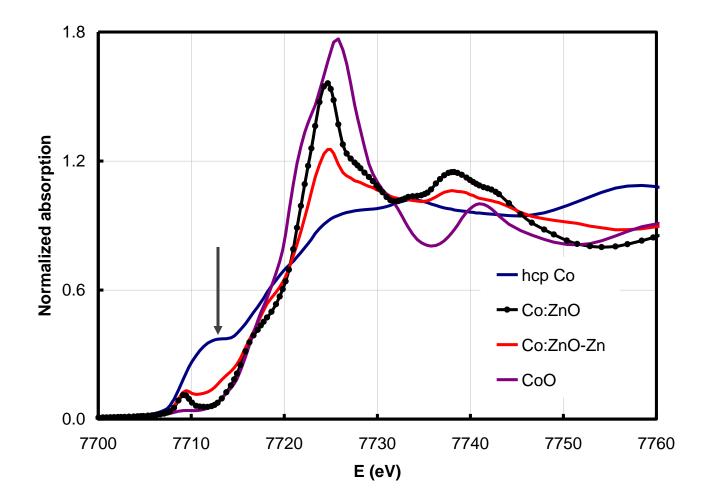
## The Co doped ZnO samples

See Kaspar etal, PRB 77, 201303 and Heald etal PRB 79, 075202

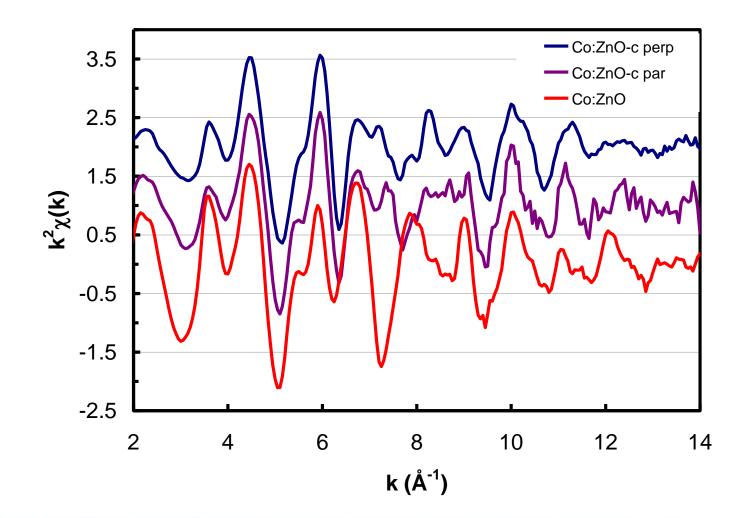
- PLD deposited samples on r- and c-plane sapphire
- 100 nm with about 16% Co
- Some samples annealed in Zn vapor to induce magnetism



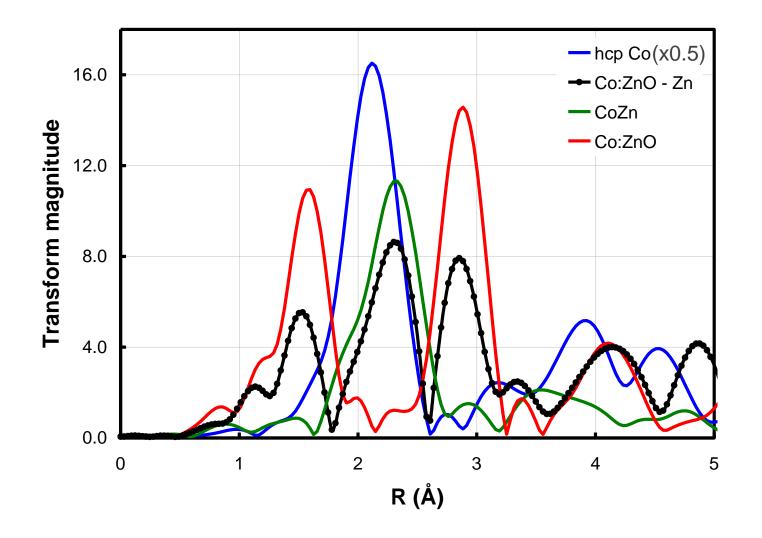
### **Detection of metallic Co - XANES**



### **EXAFS for Co doped ZnO**



#### **Detection of metallic Co - EXAFS**



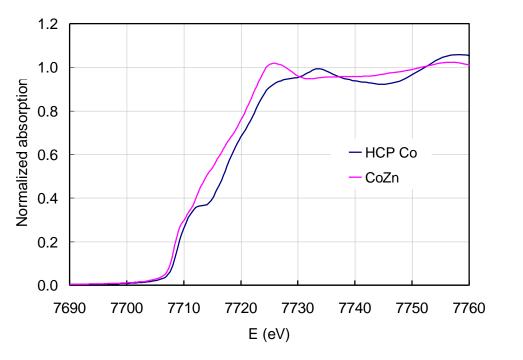
## Possible magnetic forms of Co

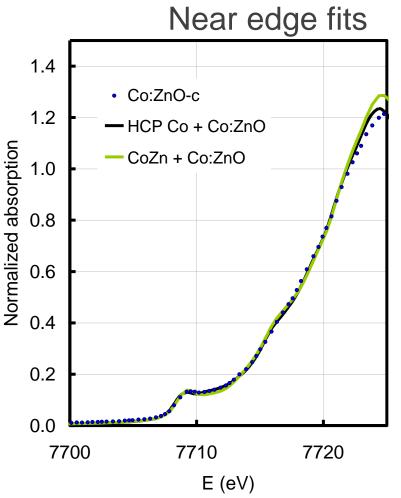
- HCP Co
  - Identified previously by XRD and XAFS
  - Typically small nanoparticles
  - 1.72  $\mu_{b}$ /Co
- FCC Co
  - Stable at high temperatures
  - Can contain up to 40% Zn with accompanying lattice expansion
  - $\,$  1.72  $\mu_{b}/\text{Co}$  for the pure phase
- CoZn intermetallic
  - β-Mn structure
  - 1.35  $\mu_b$ /Co

## Fitting of near edge

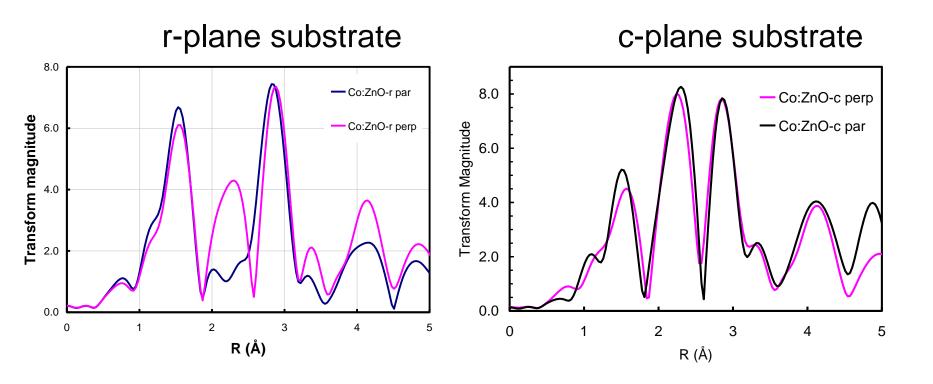
Near edge inconclusive in distinguishing the metal component

Does provide an accurate estimate of the amount of metal

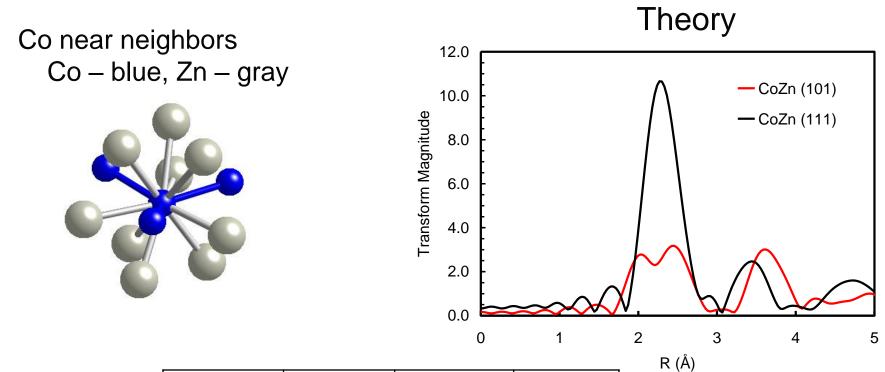




### Co EXAFS polarization dependence



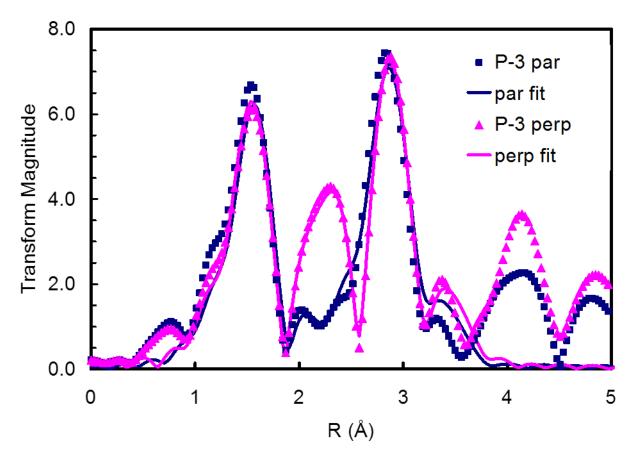
## Origin of polarization dependence - CoZn



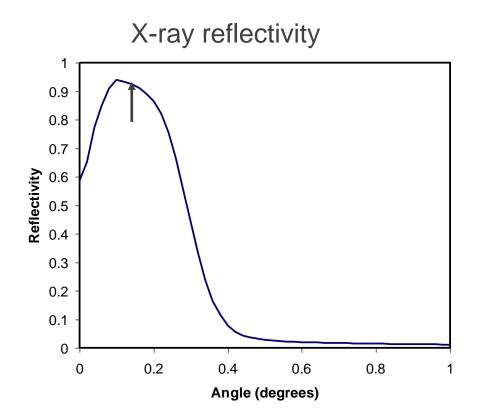
Neighbor	Ν	Distance	Measured
Со	3	2.36 Å	2.34
Zn	3	2.57	2.54
Zn	3	2.63	2.60
Zn	3	2.67	2.64

## Fitting using CoZn model

Simultaneous fit to the two polarization directions with oriented CoZn second phase



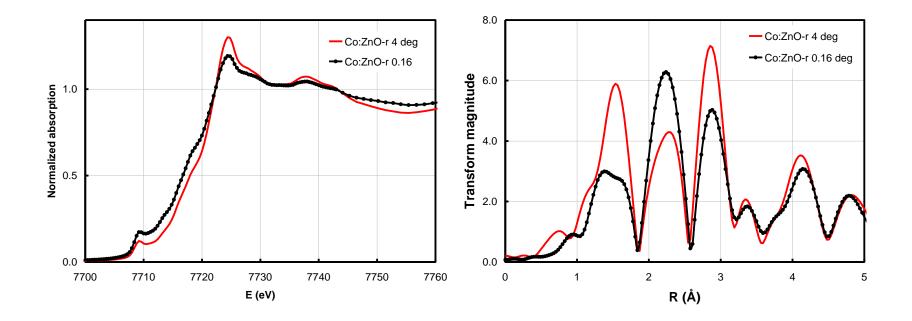
## Glancing angle XAFS - surface sensitivity



Small angles total external reflection  $\rightarrow$ x-ray penetration a few nm

Compare measurements at: 0.16 deg (surface) 4 deg (bulk)

## Angle dependent results



#### Metallic component enhanced near the surface

# X-ray Absorption Spectroscopy is a powerful tool that complements scattering methods

- Element specific
- Provides both structural (EXAFS) and chemical (XANES) information
- Fluorescence detection allows extremely dilute (ppm) components to be studied
- Does not need long range order wide range of applications
  - Operando studies of catalysts, fuel cells, and batteries
  - Detailed study of the active site in metalloproteins
  - Hazardous elements in the environment
  - Magnetism using polarized x-rays
  - High pressure studies
  - Thin film growth and characterization
  - Many others...