Introduction to Neutron and X-Ray Scattering

Sunil K. Sinha

UCSD/LANL

Acknowledgements: Prof. R. Pynn (Indiana U.)
Prof. M. Tolan (U. Dortmund)
Wilhelm Conrad Röntgen 1845-1923

1895: Discovery of X-Rays
Nobel Prizes for Research with X-Rays

1901 W. C. Röntgen in Physics for the discovery of x-rays.
1914 M. von Laue in Physics for x-ray diffraction from crystals.
1915 W. H. Bragg and W. L. Bragg in Physics for crystal structure determination.
1917 C. G. Barkla in Physics for characteristic radiation of elements.
1924 K. M. G. Siegbahn in Physics for x-ray spectroscopy.
1927 A. H. Compton in Physics for scattering of x-rays by electrons.
1936 P. Debye in Chemistry for diffraction of x-rays and electrons in gases.
1962 M. Perutz and J. Kendrew in Chemistry for the structure of hemoglobin.
1962 J. Watson, M. Wilkins, and F. Crick in Medicine for the structure of DNA.
1979 A. McLeod Cormack and G. Newbold Hounsfield in Medicine for computed axial tomography.
1988 J. Deisenhofer, R. Huber, and H. Michel in Chemistry for the structures of proteins that are crucial to photosynthesis.
2006 R. Kornberg in Chemistry for studies of the molecular basis of eukaryotic transcription.
2009 V. Ramakrishnan, T. A. Steitz and A. E. Yonath for studies of the structure and function of the ribosome.
Synchrotron research on proteins has led to major advances in drugs to battle infection, HIV, cancer.

Renal cancer drug pazopanib™ developed in part based on APS research (GlaxoSmithKline).

Close-up view of the drug binding site within HIV protease (Kaletra®, Abbott).

Ramakrishnan, Steitz and Yonath
2009 Chemistry Nobel Laureates

APS protein structure output is almost twice that of any other light source.
Designing antibiotics -
difference between bacterial and eukaryotic ribosomes is one amine group in the 2.5MD ribosome

Erythromycin – a macrolide antibiotic that blocks protein synthesis by binding to bacterial ribosomes but not to eukaryotic ribosomes

www.molgen.mpg.de
Functional domain dynamics in proteins

**NSE**
- 0.5-50 nm length scale
- ps - μs time scale
- orientational average

**FRET**
- fixed defined position
- > μs timescale
- phosphoglycerate kinase

**NMR**
- ps - ms timescale
- small proteins
Neutron and X-ray Scattering:

“small” science at big facilities!
Advantages of Neutrons and X-Rays

- Penetrating/ Non Destructive N (X)
- Right wavelength/energy N,X
- Magnetic probe N,X
- Contrast matching N
- Weakly interacting-Born approxn. N,X
- *Global* Statistical information N,X
- Buried Interfaces—depth dependence N,X
X-rays and neutrons are complementary to SPM’s

• Yield GLOBAL statistical properties about assemblies of particles
• Can be used to study BURIED interfaces or particles
• Impervious to sample environmental conditions, magnetic fields, etc.
• Can also be used to study single nanoparticles (synchrotron nanoprobe)
Nobel Prize in Physics, 1994

Awarded for “pioneering contributions to the development of neutron scattering techniques for studies of condensed matter”

Bertram N. Brockhouse
Development of neutron spectroscopy

Clifford G. Shull
Development of the neutron diffraction technique
The 1994 Nobel Prize in Physics – Shull & Brockhouse

Neutrons show where the atoms are....

When the neutrons collide with atoms in the sample material, they change direction (are scattered) – elastic scattering.

Atoms in a crystalline sample

Research reactor

Neutron beam

Detectors record the directions of the neutrons and a diffraction pattern is obtained.

The pattern shows the positions of the atoms relative to one another.

Crystal that sorts and forwards neutrons of a certain wavelength (energy) – monochromated neutrons

3-axis spectrometer with rotatable crystals and rotatable sample

3-axis spectrometer

Changes in the energy of the neutrons are first analysed in an analyser crystal...

When the neutrons penetrate the sample they start or cancel oscillations in the atoms. If the neutrons create phonons or magnons they themselves lose the energy these absorb – inelastic scattering

...and the neutrons then counted in a detector.

...and what the atoms do.
First Study of an Antiferromagnetic Structure

Antiferromagnetic Structure of MnO
(Shull and Wollan Phys. Rev. 83, 333 (1951))
Magnetic Structure of the Rare Earth Metals
(W.C. Koehler (1965))
Neutron Advantages

- Penetrating, but does no damage to sample
- H/D contrast matching can be used to study macromolecules in solution, polymers, etc.
- Strongly interacts with magnetic moments
- Energies match those of phonons, magnons, rotons, etc.
Historic accomplishments (Neutrons)

• Antiferromagnetic Structures
• Rare earth spirals and other spin structures
• Spin wave dispersion
• Our whole understanding of the details of exchange interactions in solids
• Magnetism and Superconductivity
• Phonon dispersion curves in crystals; quantum crystals and anharmonicity
• Crystal fields
• Excitations in normal liquids
• Rotons in superfluid helium
• Condensate fraction in helium
Recent Applications

- Quantum Phase Transitions and Critical points
- Magnetic order and magnetic fluctuations in the high-Tc cuprates
- Gaps and low-lying excitations (including phonons) in High-Tc
- Magnetic Order and spin fluctuations in highly-correlated systems
- Manganites
- Magnetic nanodot/antidot arrays
- Exchange bias
Recent Applications (contd.)

- Proton motion in carbon nanotubes
- Protein dynamics
- Glass transition in polymer films
- Protonation states in biological macromolecules from nuclear density maps
- Studies of protein diffusive motion in hydrated enzymes
- Boson peaks in glasses
- Phase diagrams of surfactants
- Lipid membranes
Applications in Soft Matter and Materials

- Scaling Theory of polymers
- Reptation in Polymers
- Alpha and beta relaxation in glasses
- Structures of surfactants and membranes
- Structure of Ribozome
- Excitations and Phase transitions in confined Systems (phase separation in Vycor glass; Ripplons in superfluid He films, etc.)
- Momentum Distributions
- Materials—precipitates, steels, cement, etc.
Science with X-Rays

- Diffraction and crystal structures
- Structure Factors of liquids and glasses
- Surface and Interface structures
- Structures of Thin Films
- ARPES
- EXAFS, XANES
- Studies of Magnetism with resonant XMS
- Inelastic X-ray scattering: phonons, electronic excitations
- X-ray Photon Correlation Spectroscopy
- Microscopy
- Imaging/Tomography
Applications of Surface/Interface Scattering

- study the morphology of surface and interface roughness
- wetting films
- film growth exponents
- capillary waves on liquid surfaces (polymers, microemulsions, liquid metals, etc.)
- islands on block copolymer films
- pitting corrosion
- magnetic roughness
- study the morphology of magnetic domains in magnetic films.
- Nanodot arrays
- Tribology, Adhesion, Electrodeposition
S.R. and neutron based research can help us to understand:

• How the constituent molecules self-assemble to form nanoparticles.
• How these self-organize into assemblies
• How structure and dynamics lead to function
• How emergent or collective properties arise
Why Synchrotron-radiation?

Intensity !!!
<table>
<thead>
<tr>
<th>Source Description</th>
<th>Wavelength Range (Å)</th>
<th>Δν/ν</th>
<th>Apparent Source (mm)</th>
<th>Ω (sterad)</th>
<th>Integrated Flux (W)</th>
<th>Spectral Brilliance (phot/sec/0.1% band/mrad²/mm²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sealed-off X-ray tube</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>2 kW, 50 kV</td>
<td>1.54</td>
<td>0.05%</td>
<td>1x1</td>
<td>0.1</td>
<td>0.01</td>
<td>1.6 × 10⁸</td>
</tr>
<tr>
<td>V.H. power rot.-anode tube 50 kW 50 kV</td>
<td>1.54</td>
<td>0.05%</td>
<td>1x1</td>
<td>0.1</td>
<td>0.27</td>
<td>4 × 10⁹</td>
</tr>
<tr>
<td>μ-focus rot.-anode tube 3.5 kW 50 kV</td>
<td>1.54</td>
<td>0.05%</td>
<td>0.1x0.1</td>
<td>0.1</td>
<td>0.02</td>
<td>2.8 × 10¹⁰</td>
</tr>
<tr>
<td>ACO (Orsay) (operating)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>0.54 GeV, 150 mA, B=1.6T</td>
<td>40 Å</td>
<td></td>
<td>wh. 0.5x0.6</td>
<td>2 × 10⁻⁶</td>
<td>0.16</td>
<td>4.6 × 10¹²</td>
</tr>
<tr>
<td>ADONE (Frascati) (operating)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>1.5 GeV 100 mA, B=1T</td>
<td>8.3</td>
<td></td>
<td>wh. 1x0.4</td>
<td>2 × 10⁻⁶</td>
<td>1.4</td>
<td>3.5 × 10¹³</td>
</tr>
<tr>
<td>1.8T 5-pole wiggler</td>
<td>4.6</td>
<td></td>
<td>1x0.1</td>
<td>3 × 10⁻⁶</td>
<td>2.5</td>
<td>6.9 × 10¹⁴</td>
</tr>
<tr>
<td>SRS (Daresbury) (under constr.)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>2 GeV 300 mA, B=1.2T</td>
<td>3.9</td>
<td></td>
<td></td>
<td>1/2 × 10⁻⁶</td>
<td>12</td>
<td>7.6 × 10¹³</td>
</tr>
<tr>
<td>4.5T 1 pole wiggler</td>
<td>1.1</td>
<td></td>
<td></td>
<td></td>
<td>45</td>
<td></td>
</tr>
<tr>
<td>ESRF (European S.R. Facility, proposal)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>5 GeV, 500 mA, B=0.7T</td>
<td>1</td>
<td></td>
<td></td>
<td>1/5 × 10⁻⁶</td>
<td>185</td>
<td>7.9 × 10¹⁴</td>
</tr>
<tr>
<td>3T 1 pole wiggler</td>
<td>0.023</td>
<td></td>
<td></td>
<td></td>
<td>790</td>
<td></td>
</tr>
<tr>
<td>Undulator λ=5.6 cm, B=0.2T</td>
<td>5</td>
<td></td>
<td></td>
<td>2 × 10⁻⁹</td>
<td>78</td>
<td>7.10¹⁸</td>
</tr>
<tr>
<td>5 m long (λ=5th harmonic)</td>
<td>1</td>
<td></td>
<td></td>
<td></td>
<td>26.4</td>
<td>4.7 × 10¹⁷</td>
</tr>
</tbody>
</table>
Example 1: X-Ray Diffraction & structural biology

• D.C. Phillips presents the 3-D structure of lysozyme to the Royal Society in 1965

• Linear polypeptide chain

• Folded model of the same amino acid sequence

• July 2009: 58,588 structures in Protein Data Bank

A single protein structure used to be the project of a scientific lifetime

Synchrotron Radiation - 8301 structures solved in 2009
Compare the evolution of high intensity optical and x-ray sources

High-intensity at optical wavelengths
- high harmonic generation
- tabletop coherent x-ray radiation
- attosecond pulses

High-intensity at x-ray wavelengths

G. Mourou RMP 2006

D. Moncton, George Brown
Neutron Fluxes

Inside a moderator (reactor source) $\Phi = 1.5 \times 10^{15}$ n/cm$^2$/s (steady state, e.g. ILL Grenoble, France)

Typical monochromatic flux at sample: $1.0 \times 10^{8}$ n/cm$^2$/s

In moderator of Spallation Neutron Source (e.g. SNS @ 2 MW)

$\Phi = 3.0 \times 10^{16}$ n/cm$^2$/s (Peak) $\Phi = 4.0 \times 10^{13}$ n/cm$^2$/s (Average)
Interaction Mechanisms

- Neutrons interact with atomic nuclei via very short range (~fm) forces.
- Neutrons also interact with unpaired electrons via a magnetic dipole interaction.
The Neutron has Both Particle-Like and Wave-Like Properties

- Mass: $m_n = 1.675 \times 10^{-27}$ kg
- Charge = 0; Spin = $\frac{1}{2}$
- Magnetic dipole moment: $\mu_n = -1.913 \mu_N$
- Nuclear magneton: $\mu_N = \frac{eh}{4\pi m_p} = 5.051 \times 10^{-27}$ J T$^{-1}$
- Velocity ($v$), kinetic energy ($E$), wavevector ($k$), wavelength ($\lambda$), temperature ($T$).
- $E = m_nv^2/2 = k_B T = (\frac{h k}{2\pi})^2/2m_n; \quad k = 2 \pi/\lambda = m_n v/(h/2\pi)$

<table>
<thead>
<tr>
<th></th>
<th>Energy (meV)</th>
<th>Temp (K)</th>
<th>Wavelength (nm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cold</td>
<td>0.1 – 10</td>
<td>1 – 120</td>
<td>0.4 – 3</td>
</tr>
<tr>
<td>Thermal</td>
<td>5 – 100</td>
<td>60 – 1000</td>
<td>0.1 – 0.4</td>
</tr>
<tr>
<td>Hot</td>
<td>100 – 500</td>
<td>1000 – 6000</td>
<td>0.04 – 0.1</td>
</tr>
</tbody>
</table>

$\lambda$ (nm) = $395.6 / v$ (m/s)

$E$ (meV) = 0.02072 $k^2$ (k in nm$^{-1}$)
The photon also has wave and particle properties

<table>
<thead>
<tr>
<th>E (keV)</th>
<th>λ (Å)</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.8</td>
<td>15.0</td>
</tr>
<tr>
<td>8.0</td>
<td>1.5</td>
</tr>
<tr>
<td>40.0</td>
<td>0.3</td>
</tr>
<tr>
<td>100.0</td>
<td>0.125</td>
</tr>
</tbody>
</table>
**Thermal Neutrons**

**Advantages**

1) $\lambda_n \sim$ Interatomic Spacing

2) Penetrates Bulk Matter (neutral particle)

3) Strong Contrasts Possible (e.g. H/D)

4) $E_n \sim$ Elementary Excitations (phonons, magnons, etc.)

5) Scattered Strongly by Magnetic Moments

**Disadvantages**

1) Low Brilliance of Neutron Sources-Low Resolution or Intensities; Large Samples; Low Coherence; Surfaces Difficult

2) Some Elements Strongly Absorb (e.g. Cd, Gd, B)

3) Kinematic Restriction on $Q$ for Large $E$ Transfers

4) Restricted to Excitations $\leq 100$ meV
Synchrotron X-rays

Advantages

1) $\lambda_n$ - Interatomic Spacing
2) High Brilliance of X-ray Sources - High Resolution; Small Samples; High Degree of Coherence
3) No Kinematic Restrictions (E,Q uncoupled)
4) No Restriction on Energy Transfer that Can Be Studied

Disadvantages

1) Strong Absorption for Lower Energy Photons
2) Little Contrast for Hydrocarbons or Similar Elements
3) Weak Scattering from Light Elements
4) Radiation Damage to Samples
cross section

The effective area presented by a nucleus to an incident neutron. One unit for cross section is the barn, as in “can’t hit the side of a barn!”

\[ \sigma \text{ measured in barns:} \]
\[ 1 \text{ barn} = 10^{-24} \text{ cm}^2 \]

Attenuation = \( \exp(-N\sigma t) \)
N = # of atoms/unit volume
\( t = \text{thickness} \)
\( \Phi = \text{number of incident neutrons per cm}^2 \text{ per second} \)

\( \sigma = \text{total number of neutrons scattered per second} / \Phi \)

\[
\frac{d\sigma}{d\Omega} = \frac{\text{number of neutrons scattered per second into } d\Omega}{\Phi \ d\Omega}
\]

\[
\frac{d^2\sigma}{d\Omega \ dE} = \frac{\text{number of neutrons scattered per second into } d\Omega \ & \ dE}{\Phi \ d\Omega \ dE}
\]
Scattering by a Single (fixed) Nucleus

- range of nuclear force (~ 1 fm) is << neutron wavelength so scattering is “point-like”
- energy of neutron is too small to change energy of nucleus & neutron cannot transfer KE to a fixed nucleus => scattering is elastic
- we consider only scattering far from nuclear resonances where neutron absorption is negligible

If $v$ is the velocity of the neutron (same before and after scattering), the number of neutrons passing through an area $dS$ per second after scattering is:

$$v \, dS \, |\psi_{\text{scat}}|^2 = v \, dS \, \frac{b^2}{r^2} = v \, b^2 \, d\Omega$$

Since the number of incident neutrons passing through unit area is:

$$\Phi = v |\psi_{\text{incident}}|^2 = v$$

$$\frac{d\sigma}{d\Omega} = \frac{v \, b^2 \, d\Omega}{\Phi d\Omega} = b^2$$

so $\sigma_{\text{total}} = 4\pi b^2$
Intrinsic Cross Section: Neutrons
Intrinsic Cross Section: X-Rays

\[ \vec{E}_{\text{in}} = \vec{E}_0 e^{i(k\vec{r} - \omega t)} \]

\[ E_{i,\text{rad}}(R,t) = \frac{e}{4\pi\varepsilon_0 c^2 R} a_i(t - R/c) \]

\[ a(t - R/c) = -\frac{e}{m} \alpha(\omega) \vec{E}_0 e^{i\omega R/c} e^{-i\omega t} \]

\[ \frac{E_{i,\text{rad}}(R,t)}{E_{\text{in}}} = -r_0 \alpha(\omega) \frac{e^{ikR}}{R} \cos \psi \]

Thomson Scattering Length of the Electron
(clasical electron radius):

\[ r_0 = \frac{e^2}{4\pi\varepsilon_0 mc^2} = 2.82 \times 10^{-15} \text{ m} \]
Intrinsic Cross Section: X-Rays

\[
\left( \frac{d\sigma}{d\Omega} \right)_0 \propto \omega^4
\]

Rayleigh Scattering

\[
\omega \gg \omega_r \Rightarrow \left( \frac{d\sigma}{d\Omega} \right)_0 = r_0^2 P
\]

Resonance Scattering

\[
\alpha(\omega) = \frac{\omega^2}{\omega_r^2 - \omega^2 - i\eta\omega}
\]

Thomson Scattering

\[
\left( \frac{E_{\text{rad}}(R,t)}{E_{\text{in}}} \right)^2 = \frac{r_0^2}{R^2} |\alpha(\omega)|^2 P(\psi) = \frac{|f(\Omega)|^2}{R^2}
\]

\[
\left( \frac{d\sigma}{d\Omega} \right)_0 = \frac{1}{2} (1 + \cos^2 \psi) r_0^2 |\alpha(\omega)|^2
\]
Adding up phases at the detector of the wavelets scattered from all the scattering centers in the sample:
Wave vector transfer is defined as

\[ q = k_f - k_i \]
Neutrons

Sum of scattered waves on plane II:

$$\psi_{se} = Ae^{i\phi} \sum_i \frac{b_i}{R} e^{-i\vec{q} \cdot \vec{R}_i}$$

$$\frac{d\sigma}{d\Omega} = \frac{v dS |\psi_{se}|^2}{v |A|^2 d\Omega} = \frac{vdS}{v |A|^2} \frac{|A|^2}{R^2} \frac{1}{d\Omega} \sum_{ij} b_i b_j e^{-i\vec{q} \cdot (\vec{R}_i - \vec{R}_j)}$$

$$= \sum_{ij} b_i b_j e^{-i\vec{q} \cdot (\vec{R}_i - \vec{R}_j)}$$

X-rays

$$\frac{d\sigma}{d\Omega} = r_0^2 \sum_{ij} e^{-i\vec{q} \cdot (\vec{r}_i - \vec{r}_j)} \times \left( \frac{1 + \cos^2(2\theta)}{2} \right)$$

$$\vec{r}_i \to \text{electron coordinates}$$
For neutrons, \( b_i \) depends on nucleus (isotope, spin relative to neutron (\( \uparrow \uparrow \) or \( \downarrow \uparrow \)), etc. Even for one type of atom,

\[
b_i = \langle b \rangle + \delta b_i \quad \leftarrow \text{random variable}
\]

\[
b_i b_j = \langle b \rangle^2 + \langle b \rangle \left[ \delta b_i + \delta b_j \right] + \delta b_i \delta b_j
\]

\( \text{zero zero unless } i = j \)

\[
\langle \delta b_i^2 \rangle = \langle b^2 \rangle - \langle b \rangle^2
\]

\[
\therefore \frac{d\sigma}{d\Omega} = \langle b \rangle^2 \sum_{ij} e^{-i\vec{q} \cdot (\vec{R}_i - \vec{R}_j)} + \left[ \langle b^2 \rangle - \langle b \rangle^2 \right] N
\]

\[
\sigma_{coh/4\pi} \quad \text{"coherent"} \quad \sigma_{inc/4\pi} \quad \text{"incoherent"}
\]

In most cases, we must do a thermodynamic or ensemble average

\[
\frac{d\sigma}{d\Omega} = \langle b \rangle^2 S(q) \quad S(q) = \left\langle \sum_{ij} e^{-i\vec{q} \cdot (\vec{R}_i - \vec{R}_j)} \right\rangle
\]

\[
\{R_i\} = \text{nuclear posns}
\]
x-rays
Now \( \text{Fourier Transform of particle density} \)

Proof:

So

So...
Values of $\sigma_{\text{coh}}$ and $\sigma_{\text{inc}}$

<table>
<thead>
<tr>
<th>Nuclide</th>
<th>$\sigma_{\text{coh}}$</th>
<th>$\sigma_{\text{inc}}$</th>
<th>Nuclide</th>
<th>$\sigma_{\text{coh}}$</th>
<th>$\sigma_{\text{inc}}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>$^1\text{H}$</td>
<td>1.8</td>
<td>80.2</td>
<td>V</td>
<td>0.02</td>
<td>5.0</td>
</tr>
<tr>
<td>$^2\text{H}$</td>
<td>5.6</td>
<td>2.0</td>
<td>Fe</td>
<td>11.5</td>
<td>0.4</td>
</tr>
<tr>
<td>C</td>
<td>5.6</td>
<td>0.0</td>
<td>Co</td>
<td>1.0</td>
<td>5.2</td>
</tr>
<tr>
<td>O</td>
<td>4.2</td>
<td>0.0</td>
<td>Cu</td>
<td>7.5</td>
<td>0.5</td>
</tr>
<tr>
<td>Al</td>
<td>1.5</td>
<td>0.0</td>
<td>$^{36}\text{Ar}$</td>
<td>24.9</td>
<td>0.0</td>
</tr>
</tbody>
</table>

- Difference between H and D used in experiments with soft matter (contrast variation)
- Al used for windows
- V used for sample containers in diffraction experiments and as calibration for energy resolution
- Fe and Co have nuclear cross sections similar to the values of their magnetic cross sections
- Find scattering cross sections at the NIST web site at:
  http://webster.ncnr.nist.gov/resources/n-lengths/